

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
20 June 2002 (20.06.2002)

PCT

(10) International Publication Number
WO 02/48200 A2

(51) International Patent Classification⁷: **C08F**

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(21) International Application Number: PCT/US01/47547

(22) International Filing Date:
12 December 2001 (12.12.2001)

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(25) Filing Language: English

(81) Designated States (*national*): JP, KR, US.

(26) Publication Language: English

(30) Priority Data:
60/255,540 14 December 2000 (14.12.2000) US

(84) Designated States (*regional*): European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR).

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Published:
without international search report and to be republished upon receipt of that report

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For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.



WO 02/48200 A2

(54) Title: A PROCESS FOR MAKING HIGH PURITY TRANSLUCENT PERFLUOROELASTOMER ARTICLES

(57) Abstract: Perfluoroelastomer articles comprising perfluoroelastomers having copolymerized units of tetrafluoroethylene, a perfluoro(vinyl ether), and a nitrile-containing cure site monomer are at least partially cured and then exposed in air to temperatures between 150° and 350°C for 1 to 70 hours in order to lighten the article's color and render it translucent or transparent. The resulting articles contain less than 200 ppm metals.

TITLE OF INVENTION

A PROCESS FOR MAKING HIGH PURITY TRANSLUCENT
PERFLUOROELASTOMER ARTICLES

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FIELD OF THE INVENTION

This invention relates to a process for making translucent or transparent perfluoroelastomer articles which contain very low levels of metals.

BACKGROUND OF THE INVENTION

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Perfluoroelastomers have achieved outstanding commercial success and are used in a wide variety of applications in which severe environments are encountered, in particular those end uses where exposure to high temperatures and aggressive chemicals occurs. For example, these polymers are often used in seals for aircraft engines, in semiconductor manufacturing equipment, in oil-well drilling devices, and in sealing elements for industrial equipment used at high temperatures.

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The outstanding properties of perfluoroelastomers are largely attributable to the stability and inertness of the copolymerized perfluorinated monomer units that make up the major portion of the polymer backbones in these compositions. Such monomers include tetrafluoroethylene and perfluoro(alkyl vinyl) ethers. In order to develop elastomeric properties fully, perfluoroelastomers are typically crosslinked, i.e. vulcanized. To this end, a small percentage of cure site monomer is copolymerized with the perfluorinated monomer units. Cure site monomers containing at least one nitrile group, for example perfluoro-8-cyano-5-methyl-3,6-dioxa-1-octene, are especially preferred. Such compositions are described in U.S. Patents 4,281,092; 4,394,489; 5,789,489; and 5,789,509.

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Perfluoroelastomer articles such as seals, O-rings, and valve packings are typically opaque materials which are highly filled with carbon black or metallic fillers. When exposed to plasmas in end uses such as

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semiconductor manufacturing, the polymeric component of these articles is etched away, leaving the fillers as undesirable particle contaminants. Furthermore, any metals, metal oxides or metal salts originally contained
5 in the articles are released as polymer is decomposed, thus providing a second source of contamination.

It would be desirable to have perfluoroelastomer articles which are designed to be particularly suitable for use in these harsh (i.e. plasma) environments. Such seals must also have good tensile properties and low
10 compression set resistance.

Saito, et al., in U.S. Patent 5,565,512, disclose the use of ammonium salts of organic or inorganic acids as curing agents for perfluoroelastomers. Saito's perfluoroelastomer compounds are not filled. They receive a secondary vulcanization in air. The resulting articles are
15 transparent and vary in color from amber to yellow to colorless. However, the articles have higher compression set than may be required in some end use applications.

It would be beneficial to have perfluoroelastomer articles which are translucent or transparent and lightly colored or colorless, and which
20 contain a very low amount of metals, while maintaining good tensile properties and low compression set.

SUMMARY OF THE INVENTION

The present invention is directed to a cured perfluoroelastomer article which is translucent or transparent and either lightly colored or
25 colorless. The articles contain less than 200 ppm metals and have good tensile properties and compression set. More specifically, the present invention is directed to a process for making a cured translucent or transparent perfluoroelastomer article, said process comprising:

A. mixing a perfluoroelastomer copolymer containing less than 200
30 ppm metals with at least one curing agent to form a curable composition consisting essentially of perfluoroelastomer and curative, said perfluoroelastomer comprising copolymerized

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- units of (1) tetrafluoroethylene, (2) a perfluorovinyl ether selected from the group consisting of perfluoro(alkyl vinyl) ethers, perfluoro(alkoxy vinyl) ethers, and mixtures thereof, and (3) a cure site monomer selected from the group consisting of nitrile-containing fluorinated olefins and nitrile-containing fluorinated vinyl ethers; said curing agent selected from the group consisting of diaminobisphenol AF, an organic peroxide, and a compound, other than an ammonium salt of an organic or inorganic acid, that decomposes at a temperature between 40°C and 330°C to produce ammonia;
- B. shaping said curable composition to form a perfluoroelastomer article;
- C. heating said perfluoroelastomer article to a temperature sufficient to at least partially cure said perfluoroelastomer article; and
- D. heating said at least partially cured perfluoroelastomer article in air at a temperature between 150°C and 350°C in order to render the article translucent or transparent and to lighten its color.

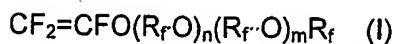
DETAILED DESCRIPTION OF THE INVENTION

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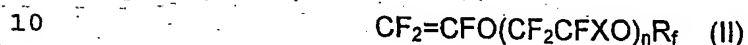
Perfluoroelastomers employed in the present invention are based on elastomeric perfluoropolymers. The perfluoroelastomers contain nitrile groups that render the polymers crosslinkable.

Perfluoroelastomers are polymeric compositions having copolymerized units of at least two principal perfluorinated monomers. Generally, one of the principal comonomers is a perfluoroolefin, while the other is a perfluorovinyl ether. Representative perfluorinated olefins include tetrafluoroethylene and hexafluoropropylene. Suitable perfluorinated vinyl ethers are those of the formula



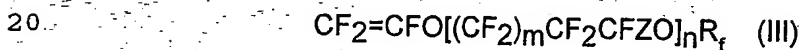
where R_f and R_f' are different linear or branched perfluoroalkylene groups
 5 of 2-6 carbon atoms, m and n are independently 0-10, and R_f is a
 perfluoroalkyl group of 1-6 carbon atoms.

A preferred class of perfluorovinyl ethers includes compositions of
 the formula



where X is F or CF_3 , n is 0-5, and R_f is a perfluoroalkyl group of 1-6
 carbon atoms.

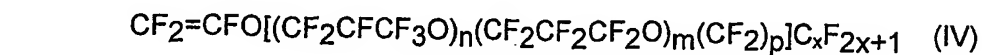
A most preferred class of perfluorovinyl ethers includes those ethers
 15 wherein n is 0 or 1 and R_f contains 1-3 carbon atoms. Examples of such
 perfluorinated ethers include perfluoro(methyl vinyl) ether and
 perfluoro(propyl vinyl) ether. Other useful monomers include compounds
 of the formula



where R_f is a perfluoroalkyl group having 1-6 carbon atoms,
 $m = 0$ or 1 , $n = 0-5$, and $Z = \text{F}$ or CF_3 .

Preferred members of this class are those in which R_f is C_3F_7 , $m = 0$, and
 25 $n = 1$.

Additional perfluorovinyl ether monomers include compounds of the
 formula

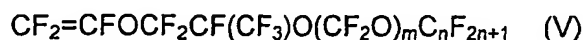


where m and n independently = 1-10, $p = 0-3$, and $x = 1-5$.

Preferred members of this class include compounds where $n = 0-1$, $m = 0-1$, and $x = 1$.

Another example of a useful perfluorovinyl ether includes

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where $n = 1-5$, $m = 1-3$, and where, preferably, $n = 1$.

Mixtures of perfluorovinyl ethers may also be used.

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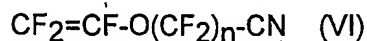
Preferred perfluoroelastomers are composed of tetrafluoroethylene and at least one perfluorovinyl ether as principal monomer units. In such copolymers, the copolymerized perfluorinated ether units constitute from about 15 mole percent to 65 mole percent (preferably 25 to 60 mole percent) of total monomer units in the polymer.

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The perfluoropolymer further contains copolymerized units of at least one nitrile group-containing cure site monomer, generally in amounts of from 0.1-5 mole percent. The range is preferably between 0.3-1.5 mole percent. Suitable cure site monomers include nitrile-containing fluorinated olefins and nitrile-containing fluorinated vinyl ethers. Useful nitrile-

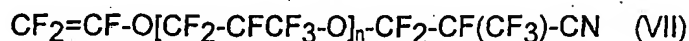
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containing cure site monomers include those of the formulas shown below.



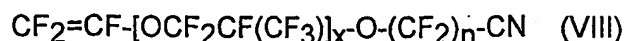
where $n = 2-12$, preferably 2-6;

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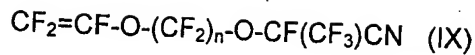


where $n = 0-4$, preferably 0-2;

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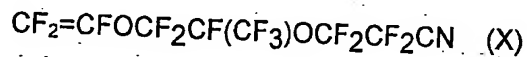


where $x = 1-2$, and $n = 1-4$; and



5 where $n = 2-4$.

Those of formula (VIII) are preferred. Especially preferred cure site monomers are perfluorinated polyethers having a nitrile group and a trifluorovinyl ether group. A most preferred cure site monomer is

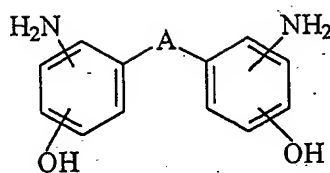


i.e. perfluoro(8-cyano-5-methyl-3,6-dioxo-1-octene) or 8-CNVE.

The perfluoroelastomers employed in this invention preferably
15 comprise copolymerized units of i) 38.5 to 74.7 (most preferably 44 to 69.5) mole percent tetrafluoroethylene (TFE), ii) 25 to 60 (most preferably 30 to 55) mole percent perfluoro(methyl vinyl) ether and iii) 0.3 to 1.5 (most preferably 0.5 to 1.0) mole percent of a nitrile group – containing cure monomer.

20 Perfluoroelastomers employed in this invention may be manufactured by such well-known processes as those described in Breazeale (U.S. Patent No. 4,281,092) or Coughlin et. al. (U.S. Patent No. 5,789,489). In order for the perfluoroelastomer articles produced by this invention to contain less than 200 ppm metals, special care must be taken
25 in order to remove sufficient impurities from the perfluoroelastomer prior to mixing it with curative. Potential impurities in the perfluoroelastomer are those resulting from the emulsion or suspension polymerization and isolation of the perfluoroelastomer copolymer. Such impurities may include various metal ions associated with pH buffers, surfactants,
30 polymerization initiators, suspension agents, defoaming agents and coagulating agents. Isolation of the perfluoroelastomer polymer from the emulsion may be improved by employing large volumes of deionized water

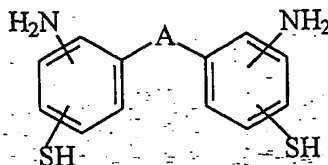
- per unit volume of emulsion when the polymer is coagulated and washed. Preferably at least 2 volumes of de-ionized water is used per volume of emulsion, most preferably at least 4 volumes of deionized water per
- 5 volume of emulsion. During isolation, the polymer emulsion is first diluted with deionized water. Coagulating agent is then added to this diluted solution. Although a variety of coagulating agents may be used during isolation (such as aluminum and calcium salts, as well as mineral acids), magnesium salts are preferred, especially magnesium sulfate.
- 10 Coagulated polymer is separated from solution by means such as filtration or centrifugation. Preferably, coagulation and washing is done at a temperature between 35°C to 65°C. The resulting perfluoroelastomer contains less than 200 ppm metals, as determined by a graphite furnace Inductively Coupled Plasma (ICP) analysis of the polymer.
- 15 Early in the process of this invention, a curable compound consisting essentially of perfluoroelastomer and a particular type of curing agent is manufactured by mixing the perfluoroelastomer with the curative. The curing agent is selected from the group consisting of 1)
- 20 bis(aminophenols), 2) bis(thioaminophenols), 3) tetraamines, 4) organic peroxides and coagents, and 5) compounds, other than an ammonium salt of an organic or inorganic acid, that decompose at a temperature between 40°C and 330°C to produce ammonia. Any type of mixing device such as a two roll rubber mill or an internal mixer (e.g. a Banbury® internal mixer or an extruder) may be employed to make the compound. By "consisting
- 25 essentially of" is meant that the compound contains substantially no fillers (i.e. less than 3 parts filler per 100 parts by weight perfluoroelastomer). By "filler" is meant common rubber additives such as carbon black, PTFE micropowders, clays and metallic fillers such as titanium dioxide, barium sulfate, and silica.
- 30 Among the curing agents mentioned above are bis(aminophenols) and bis(aminothiophenols) of the formulas



(XI)

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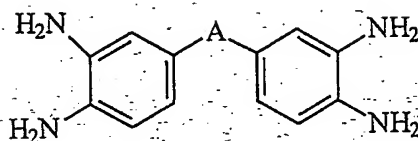
and



(XII)

and tetraamines of the formula

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(XIII)

where A is SO₂, O, CO, alkyl of 1-6 carbon atoms, perfluoroalkyl of 1-10
 15 carbon atoms, or a carbon-carbon bond linking the two aromatic rings.
 The amino and hydroxyl groups in formulas XI and XII above are
 interchangeably in the meta and para positions with respect to the group
 A. Preferably, the curing agent is a compound selected from the group
 consisting of 2,2-bis[3-amino-4-hydroxyphenyl]hexafluoropropane; 4,4'-
 20 sulfonylbis(2-aminophenol); 3,3'-diaminobenzidine; and 3,3',4,4'-
 tetraaminobenzophenone. The first of these preferred curing agents will
 be referred to as diaminobisphenol AF (DABPAF). It is an especially
 preferred curing agent. Articles prepared by the process of this invention

that are cured with DABPAF generally have better compression set (i.e. lower values) than similar articles made with a different curative. The curing agents can be prepared as disclosed in U.S. Patent Number 3,332,907 to Angelo. Diaminobisphenol AF can be prepared by nitration of 4,4'-[2,2,2-trifluoro-1-(trifluoromethyl)ethylidene]bisphenol (i.e. bisphenol AF), preferably with potassium nitrate and trifluoroacetic acid, followed by catalytic hydrogenation, preferably with ethanol as a solvent and a catalytic amount of palladium on carbon as catalyst.

Organic peroxides may also be utilized as curing agents in this invention. Useful peroxides are those which generate free radicals at curing temperatures. A dialkyl peroxide or a bis(dialkyl peroxide) which decomposes at a temperature above 50°C is especially preferred. In many cases it is preferred to use a di-tertiarybutyl peroxide having a tertiary carbon atom attached to peroxy oxygen. Among the most useful peroxides of this type are 2,5-dimethyl-2,5-di(tertiarybutylperoxy)hexyne-3 and 2,5-dimethyl-2,5-di(tertiarybutylperoxy)hexane. Other peroxides can be selected from such compounds as dicumyl peroxide, dibenzoyl peroxide, tertiarybutyl perbenzoate, and di[1,3-dimethyl-3-(t-butylperoxy)-butyl]carbonate. Generally, about 1-3 parts of peroxide per 100 parts of perfluoroelastomer are used. Another material which is usually blended with the composition as a part of the peroxide curative system is a coagent composed of a polyunsaturated compound which is capable of cooperating with the peroxide to provide a useful cure. These coagents can be added in an amount equal to 0.1 and 10 parts per hundred parts perfluoroelastomer, preferably between 2-5 parts per hundred parts perfluoroelastomer. The coagent may be one or more of the following compounds: triallyl cyanurate; triallyl isocyanurate; trimethallyl isocyanurate; tris(diallylamine)-s-triazine; triallyl phosphite; N,N-diallyl acrylamide; hexaallyl phosphoramidate; N,N,N',N'-tetraalkyl tetraphthalamide; N,N,N',N'-tetraallyl malonamide; trivinyl isocyanurate; 2,4,6-trivinyl methyltrisiloxane; and tri(5-norbornene-2-

methylene)cyanurate. Particularly useful are triallyl isocyanurate and trimethallyl isocyanurate.

Another curing agent which may be employed in this invention is a compound, other than an ammonium salt, that decomposes at temperatures between 40°C and 330°C, preferably between 90°C - 220°C, to produce ammonia. Illustrative examples of such ammonia producing compounds include aldehyde ammonia condensation products, including acetaldehyde ammonia; and other compounds, such as hexamethylenetetramine; carbamates (e.g. t-butyl carbamate, benzyl carbamate, and $\text{HCF}_2\text{CF}_2\text{CH}(\text{CH}_3)\text{OCONH}_2$); urea; urea hydrochloride; thiourea; amides (e.g. phthalamide); metal ammine complexes (e.g. tetraammine copper (II)sulfate hydrate); ammonia-Lewis acid adducts; carboxamides (e.g. oxamic acid); biuret; and unsubstituted amidines (e.g. formamidine; formamidine hydrochloride, and formamidine acetate). The preferred curatives of this type are urea, methyl carbamate and t-butyl carbamate.

In the process of this invention, the perfluoroelastomer composition prepared above is shaped into a desired article and then at least partially cured. Shaping may be performed by any conventional means such as compression molding and extrusion. Partial curing is affected by exposing the uncured article to a temperature sufficient to initiate the vulcanization reaction. Typically, an uncured article is exposed to a temperature between 150° to 240°C for a period of 5 to 30 minutes in order to at least partially cure it. The resulting article is typically deeply colored and may be opaque.

The shaped and at least partially cured fluoroelastomer article is then heated in air to a temperature between 150° and 350°C. While at this elevated temperature, additional vulcanization may occur. Also, the color of the article lightens considerably (i.e. a dark red or black article will become light amber, light yellow, or even colorless) and the article becomes either translucent or transparent to visible light. The article may

be held at this elevated temperature for any period of time, usually for a period between 1 to 70 hours (preferably 10 to 50 hours). Typically the amount of time the article is held at elevated temperature is determined by
5 the desired final color of the seal. Generally, the longer the exposure to elevated temperatures in air, the lighter the color of the resulting article. The amount and type of curative employed affects the time it is necessary to expose the article to high temperatures in order to achieve the desired color.

10 Translucent or transparent perfluoroelastomer articles made by the process of this invention have a total metal level of no more 200 ppm as measured by inductively coupled plasma (ICP). The cured articles of this invention also release less residue than do prior art perfluoroelastomer cured articles after being exposed to high temperatures or plasmas.

15 Tensile properties and compression set resistance are comparable to those of similar filled articles, e.g. tensile strengths of 6.9 to 10.3 MPa (1000-1500 psi), elongation at break of 250-300%, M_{100} of 1.7 to 2.1 MPa (250-300 psi), hardness (Shore A) of 63-72, and compression set (204°C for 70 hours) of 10-35% (preferably 10-15%), as determined by the Test
20 Methods below.

The invention is now illustrated by certain embodiments wherein all parts are by weight unless otherwise specified.

EXAMPLES

TEST METHODS

Tensile Properties

5 Physical property measurements were obtained according to methods described in ASTM D1414 (for O-rings) and ASTM D412 (for dumbbells). The following parameters were recorded:

M_{100} , stress at 100% elongation in units of MPa

T_B , tensile strength at break in units of MPa

10 E_B , elongation at break in units of %

 Compression set of O-ring samples was determined in accordance with ASTM 395-89/D1414, 204°C for 70 hours.

 The following polymers were used in the Examples:

15 Polymer A - A terpolymer comprising copolymerized units of tetrafluoroethylene (TFE), perfluoro(methyl vinyl)ether (PMVE) and 8-CNVE was prepared in the following manner. Three separate aqueous solutions were fed simultaneously to a 5.575 liter mechanically stirred, water-jacketed, stainless steel autoclave, each at a rate of 688 ml/hour.

20 Solution A contained 25.3 g ammonium persulfate, 624.5 g disodium hydrogen phosphate, and 600 g of ammonium perfluorooctanoate (Fluorad® FC-143 perfluorinated surfactant) dissolved in 20 liters of de-ionized water. Solution B contained 600 g of ammonium perfluorooctanoate dissolved in 20 liters of de-ionized water and Solution

25 C contained 25.3 g ammonium persulfate, and 600 g ammonium perfluorooctanoate dissolved in 20 liter de-ionized water. At the same time, the liquid monomer perfluoro-(8-cyano-5-methyl-3,6-dioxo-1-octene) (8-CNVE) was fed at the rate of 18.5 g/hour. By means of a diaphragm compressor, a gaseous mixture of tetrafluoroethylene (TFE) (360.1 g/hour)

30 and perfluoro(methyl vinyl) ether (PMVE) (311.3 g/hour) was fed to the reactor at a constant rate. The temperature of the reaction was maintained at 85 °C, and the pressure at 4.1 MPa (600 psi). The polymer

emulsion was removed continuously by means of a let-down valve, and the unreacted monomers were vented. A water-based silicone defoamer, type D65 available from Dow Corning Corporation, Midland, Michigan, was added continuously at the minimum rate necessary to prevent foaming of the emulsion while the gases were vented.

The emulsion from 45 hr of operation was collected and the polymer was isolated by first diluting the emulsion with 8 volumes of de-ionized water per volume of emulsion at a temperature of 60 °C. A solution of 75 g magnesium sulfate heptahydrate per liter of de-ionized water was then added to the diluted emulsion at the rate of 0.3 volume of the magnesium sulfate solution per volume of emulsion. After stirring for one hour at 60°C, the coagulated polymer was filtered, and returned to a volume of de-ionized water equal to the original dilution volume, and then stirred for 30 min at 45 °C. The polymer was again filtered and dried in an air oven for 48 hours at 70 °C. Polymer yield was 23.6 kg (51.9 lb). The composition of the polymer, as determined by FTIR, was 66.7 mole percent TFE, 32.6 mole percent PMVE, and 0.7 mole percent 8-CNVE.

Polymer B- A terpolymer containing 61.7 mole percent copolymerized units of TFE, 37.5 mole percent units of perfluoro(methyl vinyl) ether (PMVE) and 0.8 mole percent units of 8-CNVE was prepared by substantially the same procedure as was used to prepare Polymer A, with a few exceptions. The aqueous feed solutions consisted of: Solution A - 26.1 g ammonium persulfate, 686.8 g disodium hydrogen phosphate, and 600 g of ammonium perfluorooctanoate (Fluorad® FC-143 perfluorinated surfactant) dissolved in 20 liters of de-ionized water; Solution B - identical to that used for Polymer A; and Solution C - 26.1 g ammonium persulfate, and 600 g ammonium perfluorooctanoate dissolved in 20 liter de-ionized water. The 8-CNVE feed rate was 19.1 g/hr. The TFE and PMVE feed rates were 320.9 g/hour and 377.4 g/hour, respectively. The polymer was isolated by first diluting the emulsion with 5 volumes of de-

ionized water per volume of emulsion at a temperature of 40 °C. A solution of 75 g magnesium sulfate heptahydrate per liter of de-ionized water was added to the diluted emulsion at the rate of 0.3 volume of the
5 magnesium sulfate solution per volume of emulsion. After stirring for 30 min at 40°C, the coagulated polymer was filtered, and returned to a volume of de-ionized water equal to the original dilution volume, and then stirred for 30 min at 40°C. The polymer was again filtered and dried in an air oven for 48 hour at 70°C. The emulsion from 48 hr of collection yielded
10 37.4 kg (82.3 lb).

In the following Examples, the color and physical properties of various articles, made by the process this invention, were measured.

Example 1

15 Perfluoroelastomer Polymer A was compounded on a two roll rubber mill with the ingredient(s) and proportions shown in Table I. The resulting compound was molded into AS-568A K214 O-rings and press cured at 193°C for 15 minutes. The O-rings were deeply colored at this point in the process. The O-rings were then heated in air at a temperature
20 of 288°C for 24 hours. Physical properties of the resulting O-rings were measured according to the Test Methods and the results are also reported in Table I.

Example 2

25 Perfluoroelastomer Polymer A was compounded on a two roll rubber mill with the ingredient(s) and proportions shown in Table I. The resulting compound was molded into AS-568A K214 O-rings and press cured at 232°C for 30 minutes. The O-rings were then heated in air at a temperature of 316°C for 40 hours. Physical properties of the resulting O-
30 rings were measured according to the Test Methods and the results are also reported in Table I.

Example 3

Perfluoroelastomer Polymer B was compounded on a two roll rubber mill with the ingredient(s) and proportions shown in Table I. A
5 quantity of the resulting compound was molded into dumbbells and press
cured at 199°C for 10 minutes. The dumbbells were then heated in air at
a temperature of 232°C for 24 hours. Tensile properties of the resulting
dumbbells were measured according to the Test Methods and the results
are also reported in Table I. Another quantity of the above compound was
10 molded into AS-568A K214 O-rings and press cured at 199°C for 30
minutes. The O-rings were then heated in air at a temperature of 232°C for
24 hours. Compression set of the O-rings was determined according to
the Test Method and the result is included in Table I.

TABLE I

Formulation, phr	Example 1	Example 2	Example 3
Polymer A	100	100	0
Polymer B	0	0	100
DABPAF ¹	0.75	0	0
TBC ²	0	0.5	0
Lupersol 101 ³	0	0	1
Diak #8 ⁴	0	0	1
Physical Properties			
Tensile, MPa	6.7	7.1	8.7
M ₁₀₀ , MPa	1.7	1.4	1.4
Elongation, %	252	399	255
Compression set, %	12	31	31
Appearance	amber/ translucent	colorless/ transparent	amber/ translucent

¹Diaminobisphenol AF5 ²T-butyl carbamate³organic peroxide (available from Atofina)⁴tri(methallyl) isocyanurate (available from DuPont Dow Elastomers L.L.C.)

WHAT IS CLAIMED IS:

1. A process for making a cured translucent or transparent
perfluoroelastomer article having a metals content of less than 200
5 ppm, said process comprising:
 - A. mixing a perfluoroelastomer copolymer containing less than
200 ppm metals with at least one curing agent to form a
curable composition consisting essentially of
perfluoroelastomer and curative, said perfluoroelastomer
10 comprising copolymerized units of (1) tetrafluoroethylene, (2) a
perfluorovinyl ether selected from the group consisting of
perfluoro(alkyl vinyl) ethers, perfluoro(alkoxy vinyl) ethers, and
mixtures thereof, and (3) a cure site monomer selected from
the group consisting of nitrile-containing fluorinated olefins and
15 nitrile-containing fluorinated vinyl ethers; said curing agent
selected from the group consisting of diaminobisphenol AF, an
organic peroxide, and a compound, other than an ammonium
salt of an organic or inorganic acid, that decomposes at a
temperature between 40°C and 330°C to produce ammonia;
20 B. shaping said curable composition to form a perfluoroelastomer
article;
C. heating said perfluoroelastomer article to a temperature
sufficient to at least partially cure said perfluoroelastomer
article; and
25 D. heating said at least partially cured perfluoroelastomer article
in air at a temperature between 150°C and 350°C in order to
render the article translucent or transparent and to lighten its
color.

2. A process of claim 1 wherein said perfluoroelastomer employed in step A) is isolated by a process comprising:

5 i) diluting a volume of perfluoroelastomer polymer emulsion with a first amount of deionized water, said first amount equivalent to at least two volumes of emulsion, to form a dilute polymer solution;

ii) coagulating said polymer by addition of a coagulating agent to said dilute polymer solution;

10 iii) separating coagulated polymer from solution; and

iv) washing said coagulated polymer with a second amount of deionized water.

3. A process of claim 2 wherein said emulsion is diluted with at least 4 volumes of deionized water per volume of emulsion.

15 4. A process of claim 2 wherein said coagulating agent is magnesium sulfate.

20 5. A process of claim 2 wherein said coagulating and washing steps occur at a temperature between 35° to 65°C.

25 6. A process of claim 2 wherein said separating coagulated polymer from solution step is by a means selected from the group consisting of filtration and centrifugation.

7. A process of claim 2 wherein said second amount of deionized water is at least as large as said first amount for deionized water.

30 8. A process of claim 1 wherein said curing agent employed in step A) is diaminobisphenol AF.

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
20 June 2002 (20.06.2002)

PCT

(10) International Publication Number
WO 02/048200 A3

(51) International Patent Classification⁷: C08F 8/00,
214/18

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(21) International Application Number: PCT/US01/47547

(22) International Filing Date:
12 December 2001 (12.12.2001)

(74) Agent: KIRVAN, George, E.; DuPont Dow Elastomers, L.L.C., Bellevue Park Corporate Center, 300 Bellevue Parkway, Wilmington, DE 19809 (US).

(25) Filing Language: English

(81) Designated States (*national*): JP, KR, US.

(26) Publication Language: English

(84) Designated States (*regional*): European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR).

(30) Priority Data:
60/255,540 14 December 2000 (14.12.2000) US

Published:
— with international search report

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(88) Date of publication of the international search report:
3 January 2003

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(75) Inventors/Applicants (*for US only*): COUGHLIN,

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.



WO 02/048200 A3

(54) Title: A PROCESS FOR MAKING HIGH PURITY TRANSLUCENT PERFLUOROELASTOMER ARTICLES

(57) Abstract: Perfluoroelastomer articles comprising perfluoroelastomers having copolymerized units of tetrafluoroethylene, a perfluoro(vinyl ether), and a nitrile-containing cure site monomer are at least partially cured and then exposed in air to temperatures between 150° and 350°C for 1 to 70 hours in order to lighten the article's color and render it translucent or transparent. The resulting articles contain less than 200 ppm metals.

INTERNATIONAL SEARCH REPORT

Int. Application No
PCT/US 01/47547

A. CLASSIFICATION OF SUBJECT MATTER
IPC 7 C08F8/00 C08F214/18

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 7 C08F C08K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

PAJ, WPI Data, EPO-Internal

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X	WO 95 28442 A (DU PONT DE NEMOURS AND COMPANY) 26 October 1995 (1995-10-26) page 3, line 15 -page 7, line 16; claims 1-18	1,8
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☒ Patent family members are listed in annex.

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Date of the actual completion of the international search

21 August 2002

Date of mailing of the international search report

30/08/2002

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Permentier, W

INTERNATIONAL SEARCH REPORT

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PCT/US 01/47547

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